



# BioSNG and Power to Methane

New Development in Technology

September 19, 2017

Troels Stummann

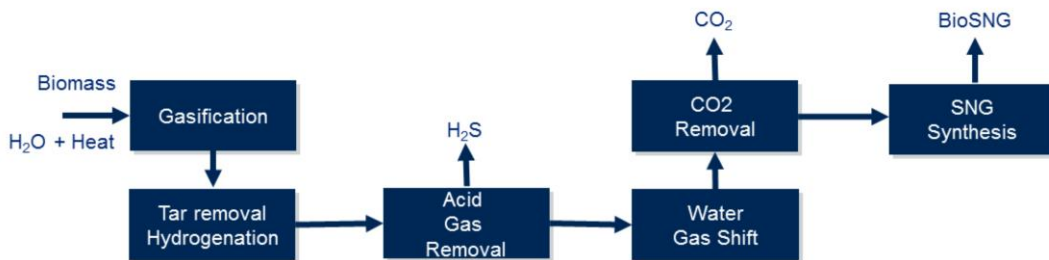
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## Agenda

- GoBiGas plant
- Power to Methane
- Boiling water reactor
- Tar reforming
- Conclusion
- Biogas upgrade Case

## GoBiGas (Göteborg, Sweden) 20 MW

- Project owner Göteborg Energy
- Gasification technology provided by Repotec/Metso
- Final tar and sulphur removal, shift and methanation technology provided by Haldor Topsøe
- Gas to pipeline 2014



3 September 19, 2017

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Topsoe has commercialized the first and only BioSNG plant in Sweden that converts 32 MW of wood pellets into 20 MW of BioSNG. Plant start-up and the first SNG to pipeline in 2014

The methanation technology employed at GoBiGas is a direct scale-down of the one used in large coal-based SNG plants (i.e. adiabatic methanators in series).

GobiGas (green building)

Tar removal; wash + TSA

2 x AGR; H<sub>2</sub>S (+CO<sub>2</sub>) and final CO<sub>2</sub> removal due to sweet shift

SNG syntheses; cascade of adiabatic reactors

While this technology is cost efficient at large scale, at smaller scale (say below 30 000 Nm<sup>3</sup>/h of product), this technology choice is too expensive. Topsoe has worked on the whole methanation section to propose a cheaper, more compact solution that cut the cost by about 30% compared to the adiabatic layout.

Power to Methane;

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## CO<sub>2</sub> removal Adjustment of gas composition

Stoichiometric for the Methanation process



$$M_E = \frac{H_2 - CO_2}{CO + CO_2} = 3.00$$

The gas composition is adjusted by removing CO<sub>2</sub>:

- Amine wash – high CAPEX and OPEX (reboiler duty)
- Rectisol/ Selexol will typical be too expensive for small scale plant
- Membrane – low efficiency

Regardless of the biomass origin, bio-syngases originating from gasification lack some hydrogen to meet the stoichiometric for the methanation reactions – which is needed to meet the required gas quality

For conventional SNG plants the gas composition are adjusted upstream  
By shift part of the CO to H<sub>2</sub> and remove CO<sub>2</sub> in an AGR unit..

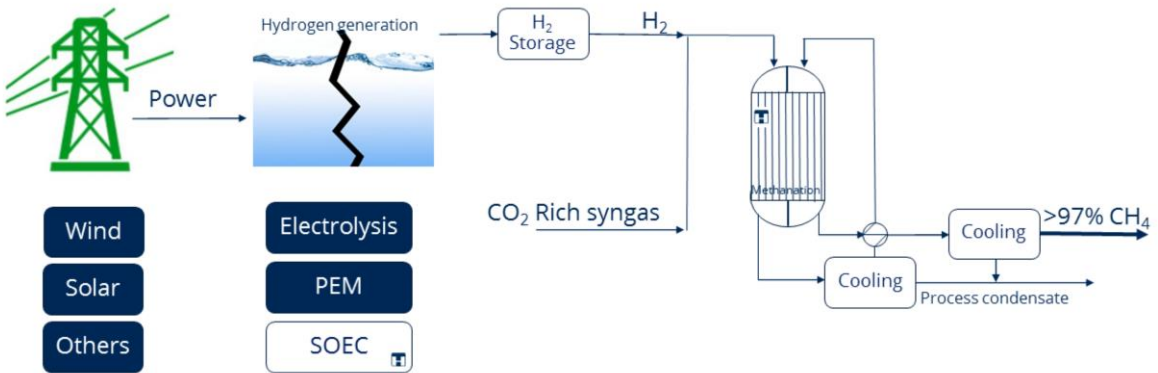
Available CO<sub>2</sub> removal technology:

Amine wash – high energy consumption to reboiler

Rectisol/ Selexol will typical be too expensive for small scale plant

Membrane – typical low efficiency (high slip of syngas)

## Power to Methane Utilise CO<sub>2</sub> as a source and not as waste



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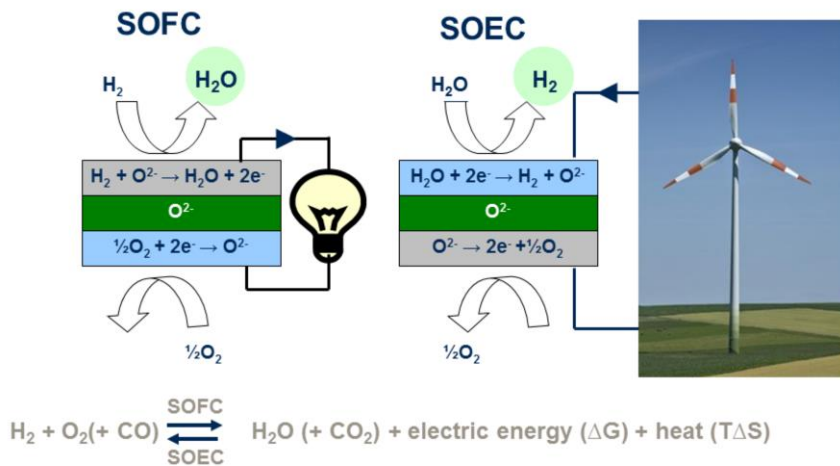
Electrolysis : alkaline  
 PEM: Polymer electrolyte membrane  
 SOEC Solid oxide electrolyser cell

Another option is to inject the required amount of hydrogen directly in the syngas thus avoiding water gas shift and the AGR unit.  
 Hydrogen can be produced by water electrolysis (alkaline / PEM / SOEC)  
 The produced oxygen can be utilized in an oxygen blown gasifier.

This approach is called Power to Methane (PtM) and can use cheap excess electricity from renewable energy sources (solar / wind) to convert it into BioSNG, that are send to the natural gas network.

Due to the availability of wind and solar power the price for power are fluctuating. By installing a H<sub>2</sub> storage as a buffer the hydrogen generation can be ramped up or down in order to utilize more power when the price is low

## Haldor Topsoe Solid Oxide Fuel Cell and Electrolyser



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SOEC Solid oxide electrolyser cell  
SOFC Solid oxide fuel cell

One option is to use SOEC which simply is a reversed SOFC  
So instead of feeding syngas and produce power, then we feed power and steam and produce  $\text{H}_2$  and  $\text{O}_2$

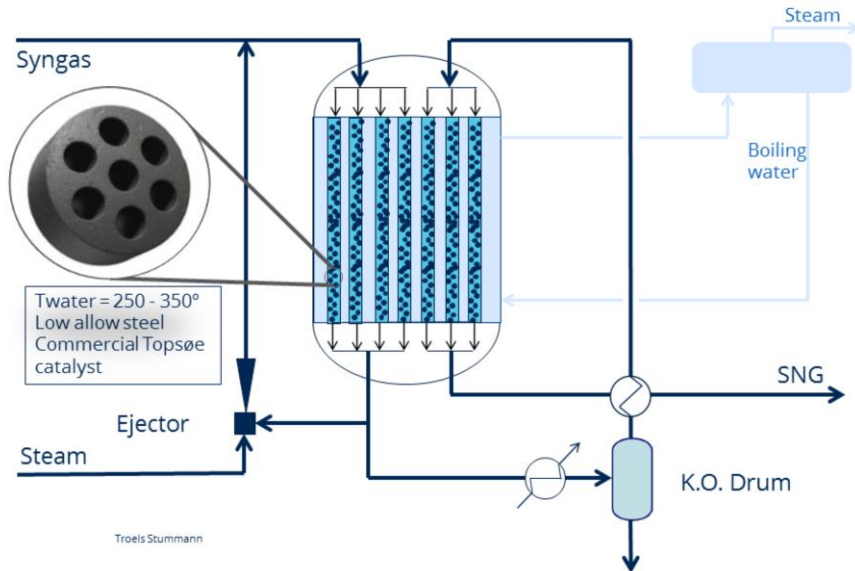
SOFC have a higher efficiency than other type of electrolyses

The boiling water reactor,  
a simpler path to methane at small scale

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## A patented boiling water reactor concept with two passes in one pressure shell



9 September 19, 2017

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The layout with adiabatic reactors used in GoBiGas can be replaced a BWR Methanation is carried out in a multitubular reactor where the heat of reaction is transferred to boiling water.

A low temperature methanation catalyst is loaded in the bottom of the tubes in order to ensure high conversion of the carbon oxides.

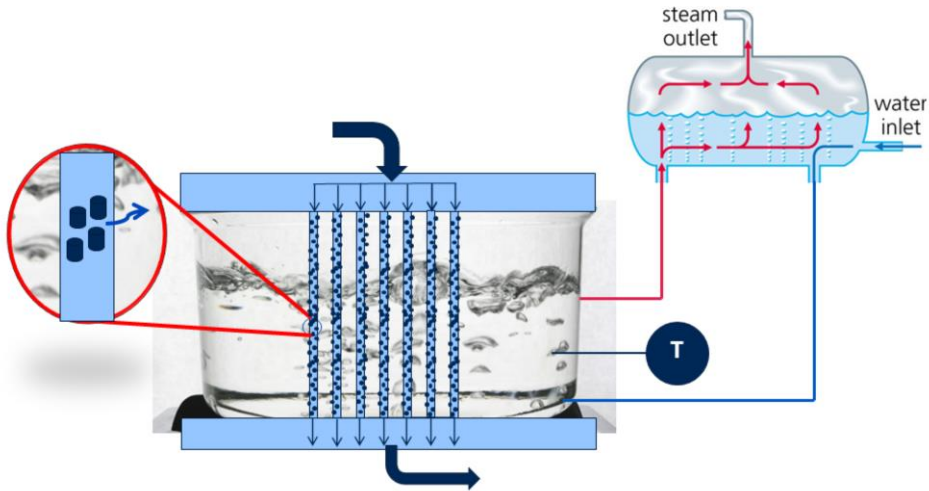
In the indicated layout we after reaching the equilibrium in the first pass and cooling down and knock out water then can have the 2<sup>nd</sup> pass, which is needed to reach the high conversion, in the same reactor.

In this way we replace 4-5 reactor, most of them refractory lined and a number of boilers and super heaters with one item.

This solution will for small size units reduce both CAPEX and space.

We expect that this cut the cost by about 30% compared to the adiabatic layout

## Boiling Water Reactor How does it work



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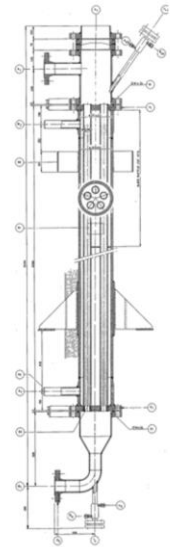
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The principle of the Boiling water reactor is simply that the catalyst is loaded in tubes which are submerged in boiling water.

This is an efficient way of removing heat from the process and controlling the temperature of all steel surfaces in contact with the water.

## Topsoe has 30 years experience with BWR

- Started with a successful BWR for SNG
- > 20 successfully running methanol BWR



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The technology comes from Topsøe methanol process and has been adapted to methanation.

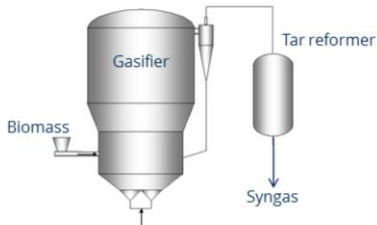
We started actually with a successful BWR pilot in Germany for SNG in 1980

Today we are mainly using the technology for MeOH - more than 20 running MeOH BWR (14 in China)

# Tar reforming

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## Tar Reforming : Convert tar to syngas



Classic operating conditions:

- 2500 ppm tar (toluene, benzene, naphthalene)
- 100 ppm S, particulates
- 850-930 °C, 1-20 bar g

Benefits:

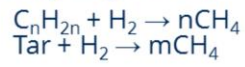
1. Increased gas production
2. Convert polyaromatic components to an extent that allows the syngas to cool for further processing without fouling or precipitation
3. Utilize the high temperature levels from gasification for increased efficiency

Other new features include a tar reformer.

Depending on the gasification technique, the layout consists of a catalytic tar reformer located directly downstream the gasifier's cyclone.

## Tar reforming reactions

Tar hydrogenation



(exothermic)  
(exothermic)

Methane reforming



(endothermic)

Requires catalytic activity  
- in the presence of sulfur !!

## Reactions

Steam reforming reactions, but in present of sulfur

The overall reaction is endotherm utilizing the high temperature outlet the gasifier.

Sulfur is needed to avoid carbon formation.

## Tar reforming - development

- Laboratory experiments at Haldor Topsøe, Denmark
- Testing at GTI Gasifier Chicago (low dust)
- In operation Skive Gasifier(Carbona), Denmark (high dust)



Have been tested at the GTI gasifier after removing the gas.  
Have been in operation at a biomass gasifier in Denmark the last 5 years operating at high dust conditions  
The photo show the monolith type version for high dust operation

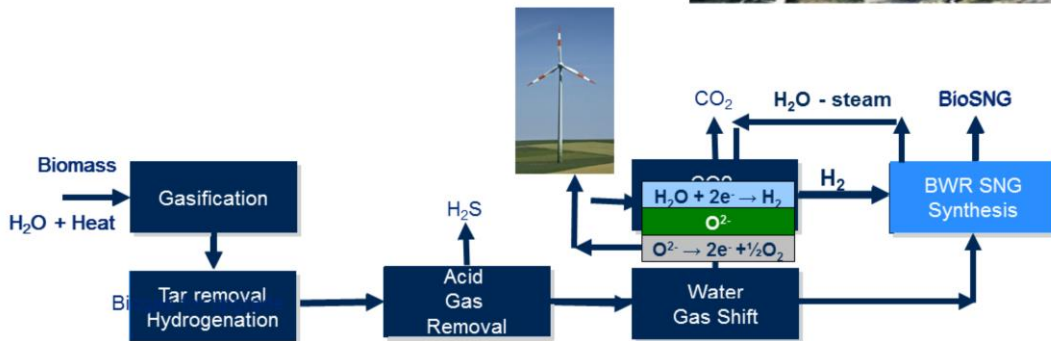
# Conclusion

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# New Development in Technology Conclusion

GoBiGas (Göteborg, Sweden)  
New layout



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The GoBiGas layout look like this.

The new layout:

- Tar reforming replace Tar removal (and hydrogenation of Olefins)
- Acid gas Removal is still needed – but for low sulfur cases a ZnO sulfur guard migh be sufficient instead of a amine wash
- Shift section will not be needed
- CO<sub>2</sub> Removal are replaced with a Hydrogen generation
- The adiabatic reactors are replaced with a BWR based SNG synthesis

This approach can be used for biogas upgrading as well.

## Biogas upgrade case – BioSNG Pilot, Foulum Denmark

10 Nm<sup>3</sup>/h = 6.2 SCFM BioGas

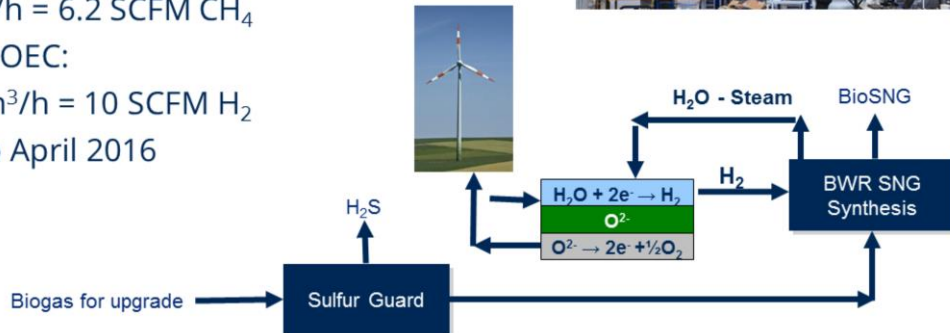
Converted to:

10 Nm<sup>3</sup>/h = 6.2 SCFM CH<sub>4</sub>

50 kW SOEC:

16.3 Nm<sup>3</sup>/h = 10 SCFM H<sub>2</sub>

Start-up April 2016



This unit was set in operation in april 2016 in Denmark for upgrading Biogas from biomass fermentation to SNG

## BioSNG Pilot, Foulum Denmark Gas compositions for biogas upgrade

	CH <sub>4</sub>	CO <sub>2</sub>	N <sub>2</sub>	H <sub>2</sub>
Inlet	56	43	1	0
Product gas	97.69	0.00	0.95	1.36

This is data from the plant: the Biogas have successful been upgraded from 56 mol% to more than 97 mole % Methane



Thank You

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